

## Corrigé exercice 2

### SOLIDES IONIQUES

cations anions	Na <sup>+</sup>	Cu <sup>2+</sup>	Zn <sup>2+</sup>	K <sup>+</sup>	Fe <sup>3+</sup>
Cl <sup>-</sup>	NaCl	CuCl <sub>2</sub>	ZnCl <sub>2</sub>	KCl	FeCl <sub>3</sub>
SO <sub>4</sub> <sup>2-</sup>	Na <sub>2</sub> (SO <sub>4</sub> )	Cu(SO <sub>4</sub> )	Zn(SO <sub>4</sub> )	K <sub>2</sub> (SO <sub>4</sub> )	Fe <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>
CO <sub>3</sub> <sup>2-</sup>	Na <sub>2</sub> (CO <sub>3</sub> )	Cu(CO <sub>3</sub> )	Zn(CO <sub>3</sub> )	K <sub>2</sub> (CO <sub>3</sub> )	Fe <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>